

A 2,3-Dihydroxygermacrene and Other Stress Metabolites of *Datura stramonium*

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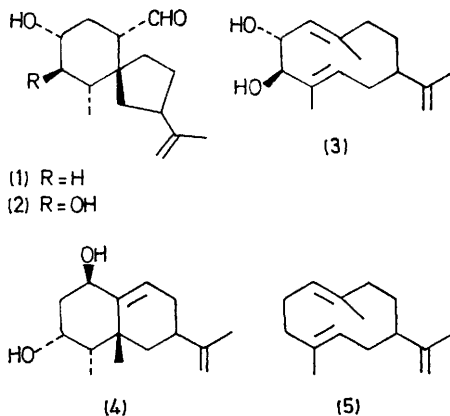
Summary A 2,3-dihydroxygermacrene (**3**) has been isolated together with lubimin, hydroxylubimin, and capsidiol from *Datura stramonium* and is of interest as a possible biogenetic precursor for several stress metabolites of the *Solanaceae*.

In a recent report¹ on the structure and distribution of the phytoalexin lubimin² (**1**), we outlined a biogenetic scheme linking most of the known, bicyclic, sesquiterpenoidal stress compounds of the *Solanaceae*. Further definition can now be added to these proposals by the isolation from *Datura stramonium* of 4-hydroxylubimin³ (**2**) and a previously unknown 2,3-dihydroxygermacrene (**3**), as stress metabolites induced together with (**1**) by inoculation with

Monilinia fructicola and other fungi. A small amount of capsidiol (**4**), already established as a stress compound from *Capsicum frutescens*⁴ and *Nicotiana* spp.,⁵ was also isolated and identified.

4-Hydroxylubimin, C₁₅H₂₄O₃ (precise mass measurement) had m.p. 96–98 °C, [α]_D²⁰ + 55° (c, 0.5 EtOH), λ_{\max} 293 (log ϵ 1.48), ν_{\max} (CHCl₃) 3585, 3420, 3090, 2745, 1725, and 900 cm⁻¹, which, with the ¹H and ¹³C n.m.r. data led to structure (**2**). The ¹³C n.m.r. spectrum in CDCl₃ contained the usual signals^{1,4} for the isopropenyl group (21.2, 109.0, and 147.0), an aldehydic carbonyl at 204.3, two –CHOH– signals at 74.1 and 76.8, three methine carbons (57.6, 47.3, and 47.1), a quaternary carbon at 48.3 and a methyl carbon at 11.2 as well as four –CH₂– groups (42.2, 32.7, 30.5, and

26.8 p.p.m.). In the ^1H n.m.r. spectrum, signals for carbinyl protons at δ 3.42 (m, v_i ca. 25 Hz) and 2.98 (dd, v_i ca. 20 Hz) were shown to arise from those protons bonded



to the carbons at 74.1 and 76.8 p.p.m. respectively, by selective decoupling. Further, the carbinyl protons were mutually coupled (J ca. 10 Hz) and the remaining coupling of ca. 10 Hz for the carbinyl absorption at δ 2.98 was eliminated by simultaneous irradiation at δ 1.58, which also collapsed the methyl doublet (δ 1.06) to a singlet. Double irradiation of the formyl proton absorption (δ 9.78, d, J ca. 3 Hz) showed this proton to be coupled with a proton at δ 2.33 whose absorption contained a large coupling of ca. 10 Hz. Other decoupling experiments indicated that the protons at δ 2.33 and 3.42 were coupled to the same two other protons. These results are accommodated by the fragment $\geq\text{C}-\text{CH}(\text{Me})-\text{CH}(\text{OH})-\text{CH}(\text{OH})-\text{CH}_2-\text{CH}(\text{CHO})-\text{C}\leq$ and the magnitudes of the vicinal coupling constants for each of the methine and carbinyl protons indicate that each is axial. Upon comparison with the data for lubimin,¹ these results clearly point to (2) as the most probable formulation.

Compound (3), $\text{C}_{15}\text{H}_{24}\text{O}_2$ (precise mass measurement) was

¹ A. Stoessl, J. B. Stothers, and E. W. B. Ward, *J.C.S. Chem. Comm.*, 1974, 709.

² L. V. Metlitskii, O. L. Ozeretskoykaya, N. S. Vul'fson, and L. I. Chalova, *Doklady Akad. Nauk S.S.S.R.*, 1971, 200, 1970.

³ During the preparation of this manuscript, N. Katsui, A. Matsunga, and T. Masamune, *Tetrahedron Letters*, 1974, 4486 reported the isolation, from infected potatoes, of a compound regarded as 'oxylubimin.' Their compound seems likely to be the same as (2) although they gave no stereochemical assignments but we have isolated (2) from inoculated potatoes.

⁴ G. I. Birnbaum, A. Stoessl, S. H. Grover, and J. B. Stothers, *Canad. J. Chem.*, 1974, 52, 993; A. Stoessl, C. H. Unwin, and E. W. B. Ward, *Phytopathol. Z.*, 1972, 74, 141.

⁵ J. A. Bailey, R. S. Burden, and G. G. Vincent, *Phytochemistry*, 1975, 14, in the press.

⁶ A. J. Weinheimer, W. W. Youngblood, P. H. Washecheck, T. K. B. Karns, and L. S. Ciereszko, *Tetrahedron Letters*, 1970, 497.

a syrup, $[\alpha]_{280}^{25} + 103$ (c, 1.0 EtOH), ν_{max} (liquid) 3370, 3080, 1670, 1650, 890, and 842 cm^{-1} , with no apparent carbonyl absorption and only end absorption in the u.v. [cf. germacrene A (5),⁶ ν_{max} (liquid) 3060, 1660, 1640, 880, and 840 cm^{-1}]. The ^{13}C n.m.r. spectrum (CDCl_3) contained the signals characteristic of the isopropenyl group (20.2, 107.7, and 153.1), four olefinic signals ($-\text{CH}=\$, 127.2 and 131.3; $=\text{C}-$, 130.9 and 140.6), two carbinyl carbon resonances ($-\text{CHOH}-$) at 74.9 and 84.4, two methyl carbons (12.5 and 17.4), one methine (51.4) and three methylene signals (34.2, 34.4, and 41.6 p.p.m.). The ^1H n.m.r. spectrum (CDCl_3) displayed three methyl singlets (δ 1.49, 1.59, and 1.73), four olefinic proton patterns and two carbinyl protons [δ 3.92 (d), J ca. 8 Hz and δ 4.35 (dd), J ca. 8 and ca. 10 Hz] which were shown to be bonded to the carbons at 84.4 and 74.9 p.p.m., respectively, by selective decoupling. $^1\text{H}-^1\text{H}$ decoupling experiments showed that the carbinyl proton at δ 4.35 was coupled to the other carbinyl proton (δ 3.92) and an olefinic proton at δ 4.95 (d), J ca. 10 Hz. These findings gave the part structure, $-(\text{Me})\text{C}=\text{CH}-\text{CH}(\text{OH})-\text{CH}(\text{OH})-\text{C}(\text{Me})=\text{CH}-\text{CH}_2-$ which, with the isopropenyl group, accounted for 12 of the 15 carbons with one methine and two methylene groups remaining. Thus, on biogenetic grounds, we propose the 2,3-dihydroxygermacrene (3) structure for this compound.

Lubimin and 4-hydroxylubimin have the same absolute configuration at C-1, as shown by their well defined Cotton effects in EtOH at 32 °C [(1), $[\alpha]_{322} + 712^\circ$, $[\alpha]_{299} 0^\circ$, $[\alpha]_{280} - 390^\circ$; (2), $[\alpha]_{322} + 1010^\circ$, $[\alpha]_{302} 0^\circ$, $[\alpha]_{279} - 820^\circ$]. This is consistent with a common origin and a hydroxygermacrene could serve as an almost direct precursor for these as well as other sesquiterpenes found in *D. stramonium* and other solanaceous plants.

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